



Natural Product Synthesis

Total Synthesis of (+)-Asperolide C by Iridium-Catalyzed Enantioselective Polyene Cyclization**

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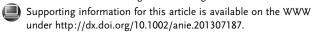
The labdane diterpenes encompass a structurally diverse class of natural products that are widely distributed in terrestrial and marine organisms.^[1] Many of these compounds exhibit notable biological properties, such as antibacterial,^[2] antimutagenic,^[3] cytotoxic,^[4] anti-inflammatory, and analgesic activities.^[5] Various procedures have been developed for their preparation,^[6] including isolation from natural sources and chemical or biological manipulation.^[7] Herein we report the first total synthesis of asperolide C (1, Scheme 1), a tetranor-

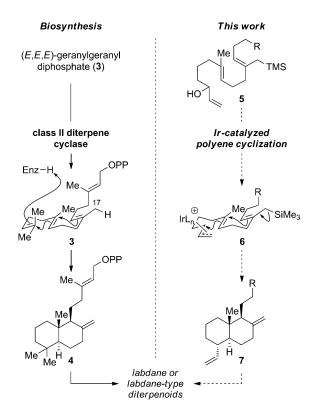
Scheme 1. Labdane skeleton, structure of asperolide C (1) and asymmetric catalytic polycyclization.

labdane diterpenoid isolated from *Aspergillus wentii* EN-48,^[8] through a unique asymmetric catalytic polycyclization cascade which is reminiscent of its biogenesis. A number of enantioselective polyolefin cyclizations have been reported in the literature.^[9] It is interesting to note, however, that applications of these processes in natural product synthesis are scarce, with only two reports emanating from the Yamamoto research group.^[10]

The carbon skeleton common to labdane-type diterpenoids is assembled biosynthetically from (E,E,E)-geranylgeranyl diphosphate (3) through a polyolefin cyclization pathway, which is consistent with the hypotheses of Stork and Eschenmoser. [11,12] The currently accepted mechanism involves protonation-initiated cyclization of 3, enabled by

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Scheme 2. Proposed biosynthesis of labdane-type diterpenoids (labda-13-en-8-yl diphosphate cation omitted for conciseness) and iridiumcatalyzed polyene cyclization.

class II diterpene cyclases, to generate a labda-13-en-8-yl diphosphate cation (Scheme 2).^[13] This intermediate typically suffers deprotonation of the C(17) methyl group to produce copalyl pyrophosphate **4**. The allylic diphosphate in **4** then engages in other rearrangement and/or cyclization reactions followed by a series of downstream biosynthetic modifications that lead to the introduction of additional functionality. It is estimated that a significant number of polycyclic diterpenoids (ca. 7000) arise from this initial cyclization event.

As part of our program on the study of iridium-catalyzed enantioselective transformations, [14] we became interested in their application to the synthesis of complex molecules. A particularly useful set of reactions that we recently disclosed includes the asymmetric cyclization of polyunsaturated allylic alcohols. [15,16] It was envisioned that this method, which employed arenes to conclude the cationic cascades, might be expanded in new ways to provide a general entry into the alicyclic labdane core. Significantly, this would require the use of an allyl silane as a terminating group. Accordingly, linear



allylic alcohol 5 (Scheme 2) would be subjected to conditions that generate π -allyl iridium complex 6. This reactive intermediate would then undergo a series of stereoselective cyclizations before suffering loss of the Me₃Si group to form the exomethylene in 7. The trans-decalin system thus obtained would be a valuable intermediate in the synthesis of various labdane or labdane-type diterpenoids.

The synthesis of asperolide C (1) commenced with the preparation of vinyl ketone 8 (Scheme 3), which was obtained from commercially available γ-butyrolactone in three steps.

Scheme 3. Reagents and conditions: a) LiHMDS (1.25 equiv), tBu- Me_2SiCl (1.25 equiv), THF/HMPA, -78 °C, d.r. > 95:5, 95%; b) PhNTf₂ (1.5 equiv), CsF (2.5 equiv), (MeOCH₂)₂, RT, d.r. > 95:5, 96%; c) 11 (1.5 equiv), $[Pd(dppf)Cl_2]\cdot CH_2Cl_2$ (10 mol%), Ph_3As (10 mol%), Cs_2CO_3 (2.5 equiv), THF/DMF/H₂O, 0°C to RT, d.r. = 10:1, 62%; d) 9-BBN (1.1 equiv), THF, 0°C to RT; then 14 (1.0 equiv), [Pd-(dppf)Cl₂]·CH₂Cl₂ (2.7 mol%), NaOH (3.0 equiv), THF/H₂O, 0°C to RT, 61%; e) PPTS (10 mol%), MeOH, RT, 81% (2 iterations). PMB = p-methoxybenzyl, TBS = tert-butylsilyl, dppf = 1,1'-bis(diphenylphosphino) ferrocene, PPTS = pyridinium p-toluenesulfonate, HMPA = hexamethylphosphoramide, HMDS = hexamethyldisilazane, Tf = trifluoromethanesulfonyl, 9-BBN = 9-borabicyclo[3.3.1]nonane.

Initial attempts to directly transform enone 8 to enol triflate 10 (LiHMDS, 2-NTf₂-5-chloropyridine, THF, -78 °C) proved unsuccessful. Therefore, a two-step protocol was examined, which involved conversion of 8 into enol silane 9 followed by exchange of the silvl group with the required triflate. Formation of 9 under standard conditions was hampered by polymerization of 8. However, the desired intermediate could be obtained in good yield (95%) and excellent Z selectivity (d.r. > 95:5) when 8 was added to a premixed solution of LiHMDS and tert-butyldimethylsilyl chloride in THF at −78°C, using HMPA as a cosolvent. Enol triflate 10 was obtained with complete retention of the olefin geometry by treatment with triflic fluoride, generated in situ following the procedure of Corey and co-workers (96%).^[17] Cross-coupling of enol triflate 10 and boronic acid 11 was achieved under the conditions of Johnson and Braun, by using [Pd(dppf)Cl₂] (10 mol %) as a catalyst in combination with Ph₃As (10 mol %) as a coligand and Cs₂CO₃ as a base to produce the desired product 12 in 62% yield (Z/E = 10:1). [18-20] The terminal olefin in diene 12 was hydroborated with 9-BBN and the resulting trialkylborane was directly subjected to B-alkyl Suzuki coupling with known vinyl iodide 14 to afford the desired polyene (61 %).[15,21] Removal of the TBS protective group required carefully chosen conditions. Whereas attempts with TBAF or para-toluenesulfonic acid led to decomposition of the material, PPTS (10 mol %) was found to catalyze the desired transformation. However, as a result of the poor stability of the product in acidic media, the reaction was usually halted at 60 % conversion. [22] Re-subjection of the recovered starting material led to allylic alcohol 13 in 81 % combined vield.

With polyene precursor 13 in hand, the pivotal iridiumcatalyzed cyclization cascade was examined (Scheme 4). Gratifyingly, reaction of 13 under standard conditions with $[{Ir(cod)Cl}_2]$ (3.2 mol%) and (R)-16 (12.8 mol%) as catalyst precursors, along with Zn(OTf)₂ (16 mol%) as a Lewis acid delivered decalin 15 with excellent stereoselectivity (d.r. = 9:1, e.r. = 98:2) and in 73 % yield. $[^{15,23}]$

Scheme 4. Iridium-catalyzed enantioselective polyene cyclization cascade

The synthesis of asperolide C (1) continued with deprotection of 15 followed by stepwise oxidation of the liberated primary hydroxy group. The carboxylic acid thus obtained was treated with trimethylsilyldiazomethane to afford the corresponding methyl ester (62% over 4 steps, Scheme 5). Epoxidation of the exomethylene group with freshly prepared DMDO at -20°C proceeded from the sterically more accessible α face to deliver oxirane 17 in moderate yield (45%, 66% brsm). Exposure of 17 to trifluoroacetic acid in anhydrous CH₂Cl₂ at 0°C led to selective epoxide opening and efficient cyclization to furnish lactone 18 (70%).[24]

With the tricyclic scaffold of the target constructed, the final stages of the total synthesis were addressed. After masking the primary hydroxy group in 18 as a TBS ether (TBSCl, imidazole, DMAP, 89%), Lemieux-Johnson oxidation afforded aldehyde 19 in 81% yield. Introduction of the quaternary center at C(4) by enolate alkylation was complicated by the presence of a γ -lactone (p $K_a \approx 20$). We reasoned that the direct alkylation of an aldehyde (p $K_a \approx 17$) could offer the necessary chemoselectivity in the deprotonation event. In the experiment, treatment of a solution of 19 in THF at -20°C with tBuOK (1.25 equiv), [25] followed by addition of iodomethane (1.25 equiv) and warming to 0 °C delivered 20 as a single isolable product. Pinnick oxidation of aldehyde 20 to the corresponding carboxylic acid (76%) and cleavage of the TBS group (74%) completed the first total synthesis of asperolide C (1). The ¹H and ¹³C NMR spectra of the synthetic material were in agreement with those reported



Scheme 5. Reagents and conditions: a) DDQ (1.1 equiv), pH 7 buffer, CH_2CI_2 , RT, 98%; b) DMP (1.5 equiv), CH_2CI_2 , RT, 80%; c) NaClO₂ (4.0 equiv), NaH_2PO_4 (6.0 equiv), 2-methyl-2-butene (70 equiv), $tBuOH/H_2O$, RT; then Me_3SiCHN_2 (1.1 equiv), $tBuOH/H_2O$, RT; then $tBuOH/H_2O$, RT; then $tBuOH/H_2O$, RT; then $tBuOH/H_2O$, $tBuOH/H_2O$, $tBuOH/H_2O$, tC, tC

for the natural product. It should be noted that asperolide C (1) was originally isolated as an inseparable 3:4 mixture with the known terpene botryosphaerin, which precluded thorough characterization. With a pure sample of 1 in hand, we could measure its optical rotation for the first time ($[\alpha]_D^{26}$ = +2.5 (c = 0.25, MeOH)).

In conclusion, the first total synthesis of the tetranorlabdane diterpenoid asperolide C (1) has been achieved. This study represents a rare example of the use of an enantioselective polyene cyclization reaction in a natural product synthesis and the first that strategically relies on modern iridium catalysis to construct the carbobicyclic core scaffold. Additionally, the described route features a series of crosscoupling reactions to efficiently assemble the linear polyene precursor. Specifically, the Pd-mediated coupling of a dienol triflate with Me₃SiCH₂B(OH)₂ provides a novel access route to allylic silanes. Moreover, a chemo- and diastereoselective alkylation of an aldehyde enolate was employed to complete the target structure. The synthetic strategy provides a general entry into the labdane-type diterpenoids and has significant potential for enabling the total synthesis of other terpene natural products.

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